

BtCH₂TMS-Assisted Homologation of Carboxylic Acids: A Safe Alternative to the Arndt–Eistert Reaction

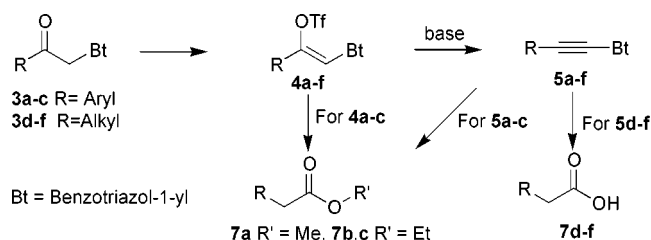
Alan R. Katritzky,* Suoming Zhang, and Yunfeng Fang

Center for Heterocyclic Compounds, Department of Chemistry, University of Florida, Gainesville, Florida 32611-7200

katritzky@chem.ufl.edu

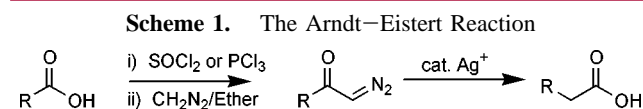
Received August 23, 2000

ABSTRACT



One-carbon homologation of carboxylic acids is achieved by (i) treatment of an acyl chloride with 1-[(trimethylsilyl)methyl]-1*H*-1,2,3-benzotriazole (BtCH₂TMS) (**1**) to afford *N*-(acylmethyl)benzotriazoles **3a–f**, followed by (ii) conversion of **3a–f** with triflic anhydride into RC(OTf)=CHBt **4a–f**, and (iii) the subsequent reaction of **4a–c** with NaOCH₃ followed by 1*N* HCl to afford esters RCH₂CO₂R' **7a–c** in overall yields of 50–70%. For the aliphatic compounds **5d–f**, treatment of **5d–f** with *p*-toluenesulfonic acid followed by TBAF/THF afforded acids RCH₂COOH **7d–f**.

The Arndt–Eistert reaction is the most important and most commonly used procedure for converting a carboxylic acid into its higher homologue acid or acid derivative, such as an ester or amide. An intermediate α -diazomethyl ketone undergoes the Wolff rearrangement in the presence of silver oxide or silver benzoate as a catalyst (Scheme 1).



The Arndt–Eistert reaction is widely used in organic synthesis.¹ However, this classical procedure suffers from certain handling difficulties and limitations for large scale preparation. α -Diazomethyl ketones are hazardous and are

strong skin irritants.² Moreover, to achieve high overall yields, the intermediate diazo compounds often have to be recrystallized and freshly prepared silver benzoate is required. Modified Arndt–Eistert procedures ease these difficulties only in part.³ Alternative one-carbon homologations of carboxylic acids include (i) Kowalsky's CH₂Br₂ method,⁴ which requires the use of a strong base and gives 53–80% yields and (ii) Barton's radical homologation,⁵ where light-sensitive *O*-acyl-*N*-hydroxyl-2-thiopyridones are the key intermediates.

In this letter, we report a convenient and safe alternative for acid homologation with the one-carbon synthon BtCH₂TMS (**1**), utilizing the anion stabilizing and leaving group properties of benzotriazole.⁶ BtCH₂TMS (**1**) and *N*-(acyl-

(2) Lee, V.; Newman, M. S. *Org. Synth.* **1988**, 613.

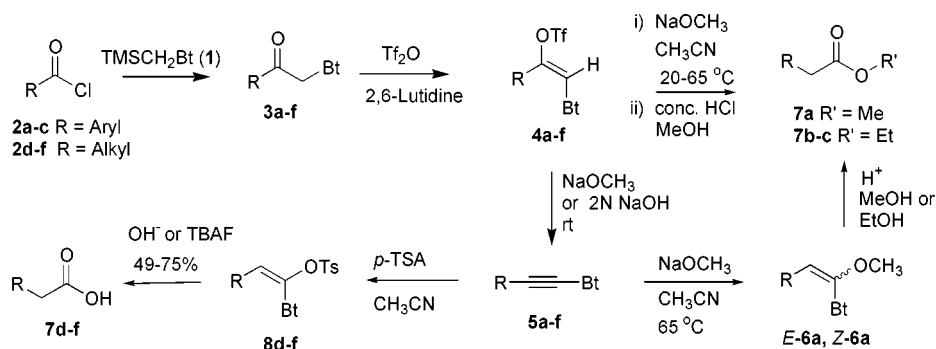
(3) (a) Winum, J.-Y.; Kamal, M.; Leydet, A.; Roque, J.-P.; Montero, J.-L. *Tetrahedron Lett.* **1996**, 1781. (b) Aller, E.; Molina, P.; Lorenzo, A. *Synlett* **2000**, 4, 526.

(4) (a) Kowalski, C. J.; Haque, M. S.; Fields, K. W. *J. Am. Chem. Soc.* **1985**, 107, 1429. (b) Reddy, R. E.; Kowalski, C. J. *Org. Synth.* **1992**, 146.

(5) (a) Barton, D. H. R.; Chern, C.-Y.; Jaszberenyi, J. C. *Tetrahedron Lett.* **1992**, 5013. (b) Barton, D. H. R.; Chern, C.-Y.; Jaszberenyi, J. C. *Tetrahedron Lett.* **1991**, 3309.

(1) Mulzer, J. In *Comprehensive Organic Functional Group Transformations*; Katritzky, A. R., Meth-Cohn, O., Rees, C. W., Eds.; Pergamon Press: Oxford, 1995; Vol. 5, pp 144 and 276.

Scheme 2



methyl)benzotriazoles **3a–f** were prepared according to the literature procedures in good yields.⁷ *N*-(Acylmethyl)benzotriazoles **3a–f** were each treated with triflic anhydride (Tf₂O) in the presence of 2,6-lutidine at 0–20 °C to afford stereospecifically *E*-enol triflates **4a–f** in excellent isolated yields (>90%, see Table 1 and Scheme 2). No *trans*-isomers

Table 1. One-carbon Homologation of Carboxylic Acids 2 to Esters 7

R	isolated yields, %			
	3	4	5	7
2a C ₆ H ₅	85 ⁷	95	98 ⁹	89 ^a
2b <i>p</i> -ClC ₆ H ₄	90	95	97	98 ^a
2c <i>p</i> -CH ₃ C ₆ H ₄	91 ⁷	90	95 ⁹	92 ^a
2d CH ₃	83 ⁷	88	90	55 ^b
2e (CH ₃) ₃ CCH ₂	95	95	91	45 ^b
2f <i>n</i> -CH ₃ (CH ₂) ₆	95	90	92	49 ^b

^a Isolated yield from **4**. ^b Yield from **5**.

were detected in the NMR spectra of the crude products: for instance, irradiation of the vinylic proton of **4a** at 7.59 ppm showed no NOE between the phenyl group and the vinylic proton, indicating that they are mutually *trans*.

Enol triflates **4a–f** reacted with 2.5 equiv of NaOCH₃ in acetonitrile at 20 °C to give alkyne-1*H*-1,2,3-benzotriazoles **5a–f** which could be isolated in quantitative yields. The effects of the base and solvent are small: use of 2 N NaOH/THF or NaOCH₃/CH₃CN gave similar results. For aromatic compounds **4a–c**, further reflux of the reaction mixture at 65 °C followed by hydrolysis with concentrated HCl in an alcohol afforded directly the corresponding homologated esters **7a–c**. The reaction sequence is shown in Scheme 2. Treatment of **5a** with 1.2 equiv of NaOCH₃ in acetonitrile at 65 °C for 2 h regioselectively generated 1-[(*E*)-1-methoxy-2-phenylethenyl]-1*H*-1,2,3-benzotriazole **E-6a** (R = Ph)⁸ as the major product and 1-[(*Z*)-1-methoxy-2-phenylethenyl]-

1*H*-1,2,3-benzotriazole **Z-6a** (R = Ph)⁸ as the minor product. No regioisomers were detected. NOE experiments of **E-6a** displayed positive NOE between OCH₃ group and vinylic proton when the vinylic proton at 6.11 ppm was irradiated. The ratio of *E*/*Z-6a* is from 74:26 to 80:20.

The mixtures of **E-6a,b** and **Z-6a,b** were hydrolyzed with concentrated HCl in methanol or ethanol yielding the homologated methyl ester **7a** or ethyl ester **7b** in good yields, respectively. In fact, the enol triflates **4a–c** readily converted into the desired esters **7a–c** after treatment in situ with NaOCH₃ in acetonitrile followed by concentrated HCl in an alcohol successively in excellent yields without the isolation of the intermediates **5a–c** and **6a–c** (Table 1).

The procedure was modified for the aliphatic derivatives **4d–f**: these compounds were treated with 2.2 equiv of NaOCH₃ in CH₃CN (or with 2N NaOH in THF) at 20 °C for 1 h. After workup, 1-alkynylbenzotriazoles (**5d–f**) were obtained in quantitative yield (Table 1, entries 4–6). This is the first synthesis of an 1-alkynylbenzotriazole, compounds that cannot be obtained by direct alkylation of benzotriazole.⁹ However, after further reflux of the reaction mixture, none of the desired addition product analogous to **6a–c** was detected. Fortunately, treatment of **5d–f** with *p*-toluenesulfonic acid monohydrate in acetonitrile at 65 °C for 4–6 h generated enol toluenesulfonates **8d–f** in 62–75% yield together with about 5–10% of the starting **3d–f** and 5–10% of the hydrolyzed product **7d–f**. Hydrolysis of isolated **8d–f** with 1 equiv of TBAF in THF at 70 °C provided the corresponding acids **7d–f** (65–80%). This two-step procedure could be carried out in one-pot by direct treatment of **5d–f** with *p*-toluenesulfonic acid followed by the hydrolysis of **8d–f** with TBAF in THF without the isolation of the intermediate **8d–f**, but gave a lower yields of **7d–f** (40–55%).

In summary, a concise and practical procedure for the transformation of acids into one-carbon homologated acid derivatives has been developed using readily available, versatile, and high-yielding reagent BtCH₂TMS (**1**), and this procedure is apparently applicable for both aromatic and aliphatic carboxylic acid derivatives. This novel approach

(6) Katritzky, A. R.; Lan, X.; Yang, J. Z.; Denisko, O. V. *Chem. Rev.* **1998**, *98*, 409.

(7) Katritzky, A. R.; Lam, J. N. *Heteroatom Chem.* **1990**, *21*.

(8) Katritzky, A. R.; Zhao, X.; Shcherbakova, I. V. *J. Chem. Soc., Perkin Trans. 1* **1991**, 3295.

(9) (a) Kitamura, T.; Tashi, N.; Tsuda, K.; Fujiwara, Y. *Tetrahedron Lett.* **1998**, 3787. (b) Kitamura, T.; Tashi, N.; Tsuda, K.; Chen, H.; Fujiwara, Y. *Heterocycles* **2000**, *52*, 303.

extends the scope of utilization of benzotriazole compounds and of homologation chemistry.

Supporting Information Available: Experimental procedures and characterization for compounds **3b,e,f**, **4a-f**,

5a-f, *E,Z*-**6a** and **8d-f**, ¹H NMR, and ¹³C NMR spectra for **3e**, **4a-f**, **5b,d-f**, *E,Z*-**6a**, and **8d-f**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL0002370